

Fluorescence, Phosphorescence et autres Luminescences

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Plan

- Ce que l'on voit
- Introduction historique
- Diagrammes d'énergie
- Zoologie
- La cinétique de l'état excité
- Conclusion

Applications

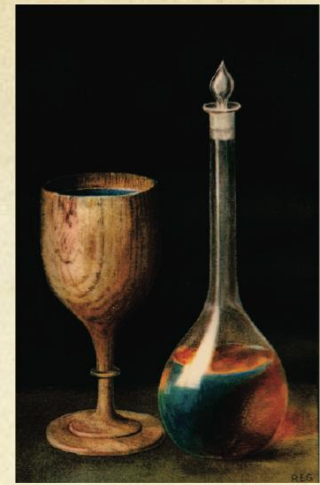
- Les azurants,
- Photovoltaïque, Thermoluminescence,
- Microscopie, Indicateurs, super résolution, Lasers
- L'acte chimique élémentaire, Oxymètres

Sources en français

- (2002)(2012) B. Valeur and M. r. N. Berberan-Santos. *Molecular Fluorescence: Principles and Applications* Ed. Wiley-VCH Verlag GmbH & Co. KGaA.
- (2007) T.-H. Tran-Thi, P. Chazette, R. Dagnélie, J. Kasparian, I. Leray, L. Magne, H. Paolacci and B. Valeur. *Détecter les polluants dans l'air et dans l'eau* l'actualité chimique. mai-juin 308-309.
- (2005) B. Valeur. *Lumière et luminescence* Ed. Belin.
- (2004) B. Valeur. *Invitation à la fluorescence moléculaire* Ed. De Boeck. 202.
- N°308-309 mai-juin 2007 *N° thématique : La photochimie pour mieux vivre*

Historique

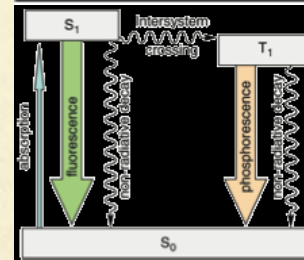
Monardes N. Dos Libros, Seville,
Spain, 1569.



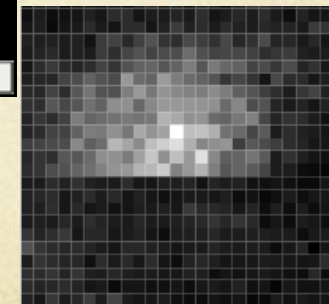
Stokes, G. G. Philos. Trans.
1852, 142, 463–562.



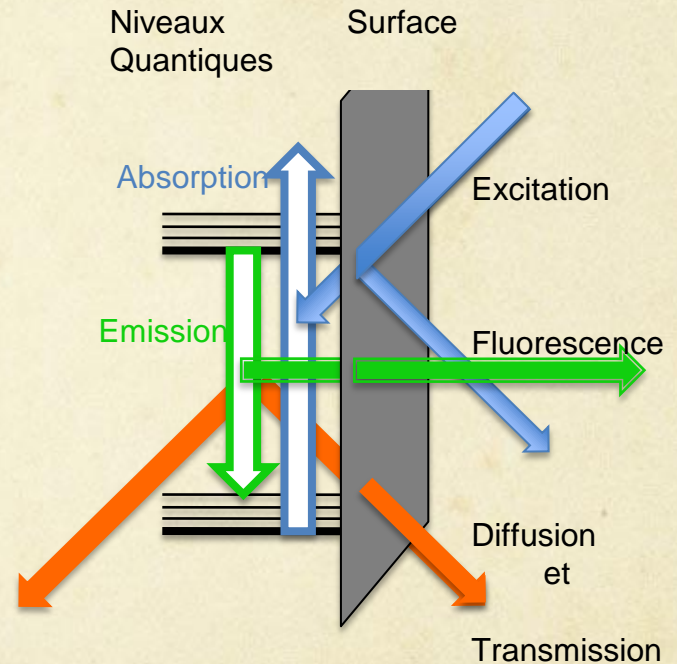
Perrin, F. Ann. Phys. (Paris, Fr.)
1929, 12, 169–275.



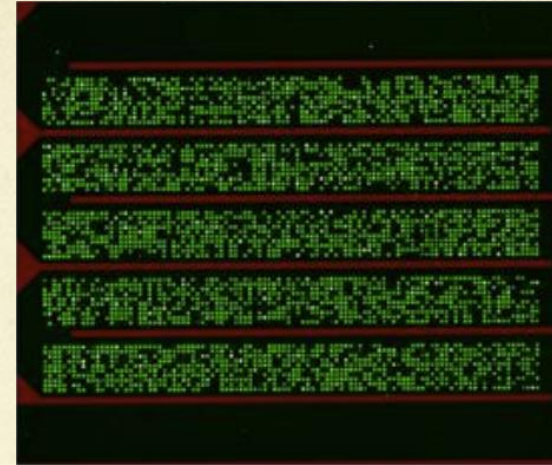
Orrit, M.; Bernard, J.
Phys.Rev.Lett. 1990, 65, 2716.



Ce que l'on voit



Autour de nous

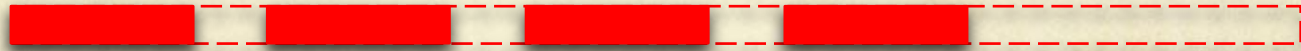


Zoologie et diagramme d'énergie

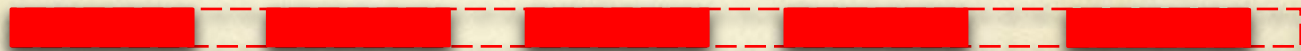
Bande de
Conduction



LUMO



HOMO



Bande de
Valence

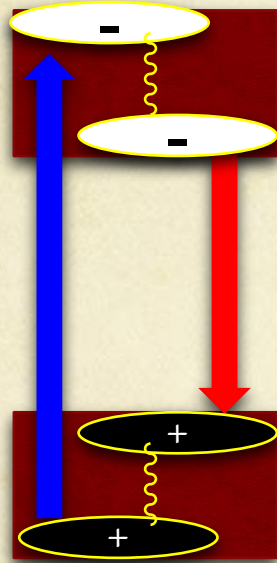


Semi Conducteurs	Défauts dans les Semi Conducteurs	Impureté dans les minéraux	Molécules fluorescentes en solution	Ions des métaux de transition	Lanthanides +3
CdSe CdSe/ZnS	ZnO	Ni, NV/Diaman t	Fluorescéine Fl. Protéines	Ru ⁺³ , Ir ⁺³ UO ₂ ⁺²	Tb, Eu, Dy, Nd, Er, Yb

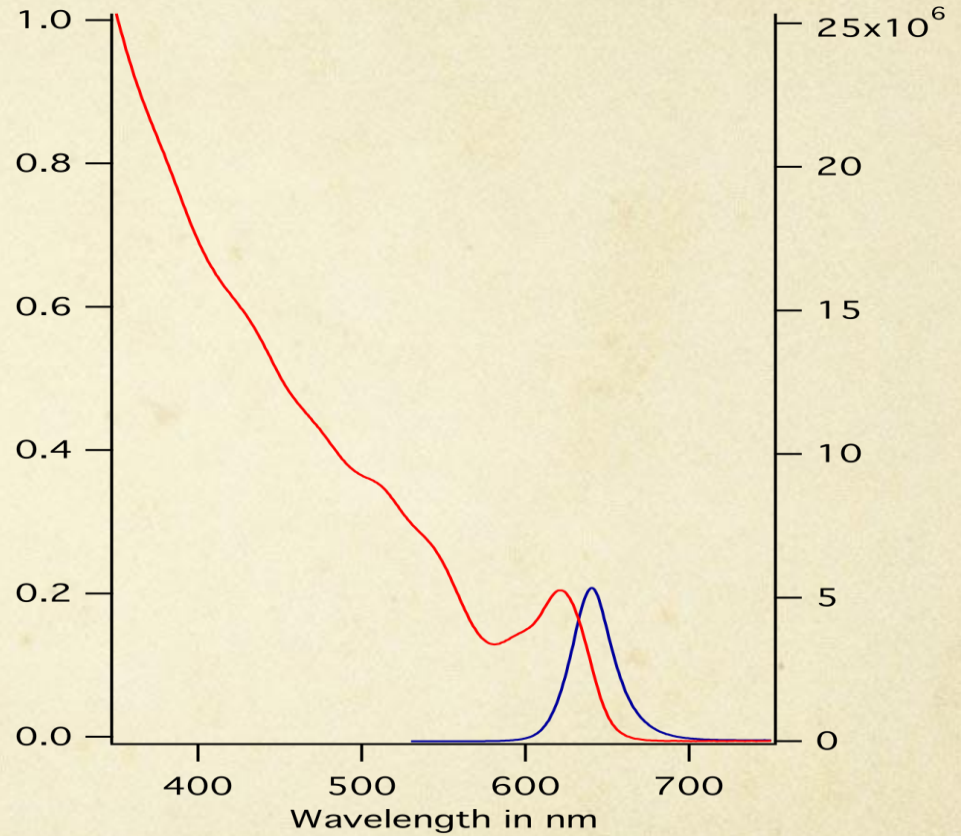
Semi-conducteurs

Bande de
Conduction

Bande de Valence



Absorbance



Fluorescence



Résumé

- Photoluminescence :
 - Excitation d'un seul électron
 - Relaxation dissipative
 - Equilibre thermodynamique de l'état excité
 - Emission de photon en compétition avec les désactivations non radiatives
 - Déplacement de stokes
 - Principe de Jablonski
 - Réactivité des états excités
 - Rendement de fluorescence

Diode Electro Luminescente

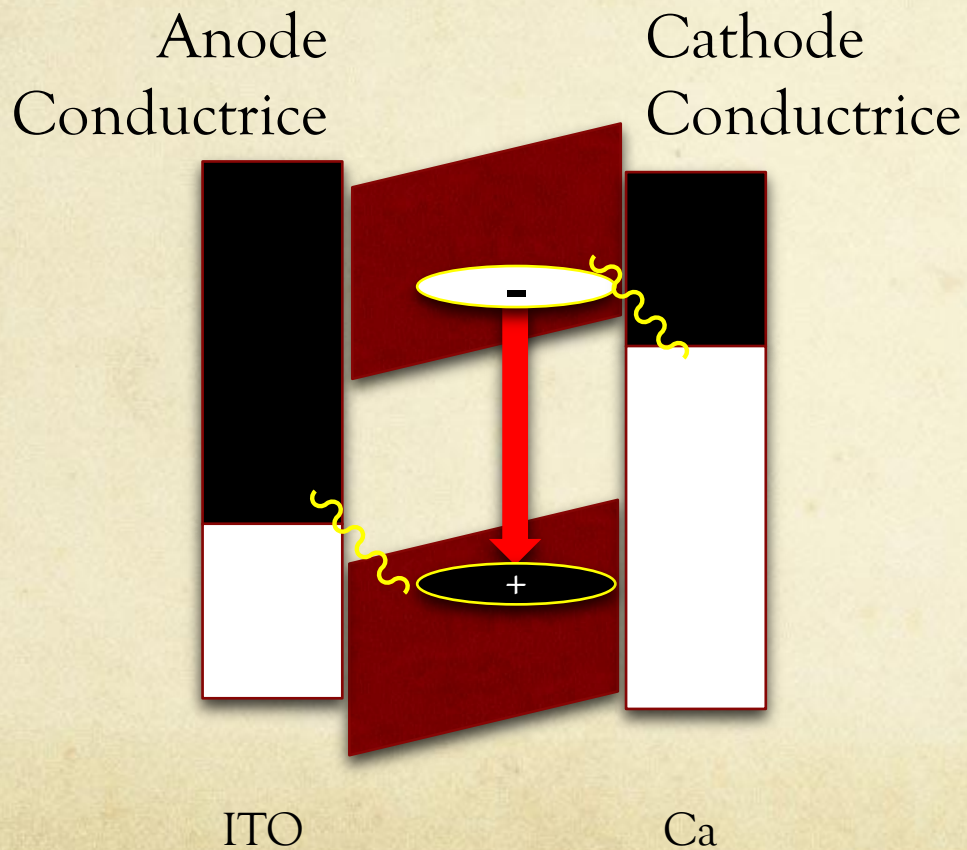
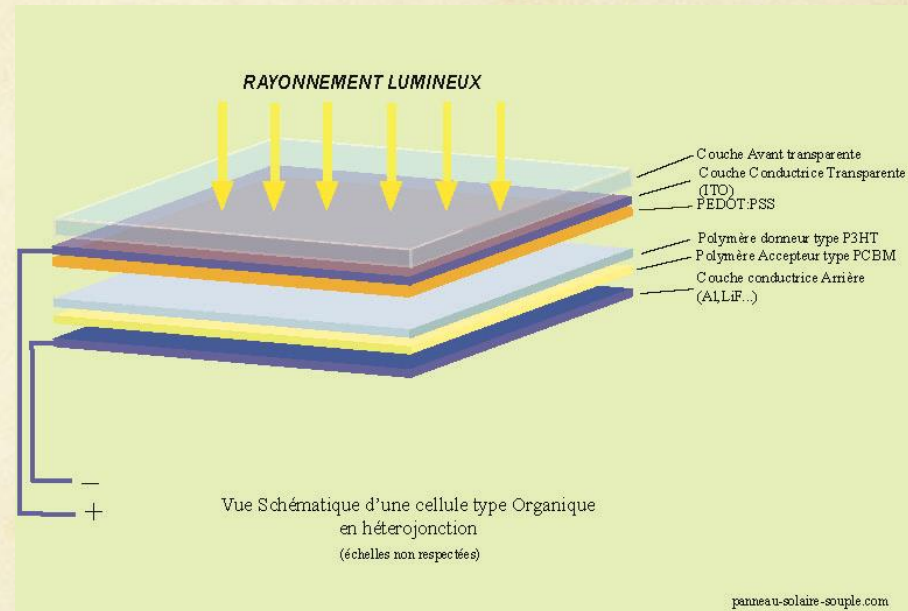
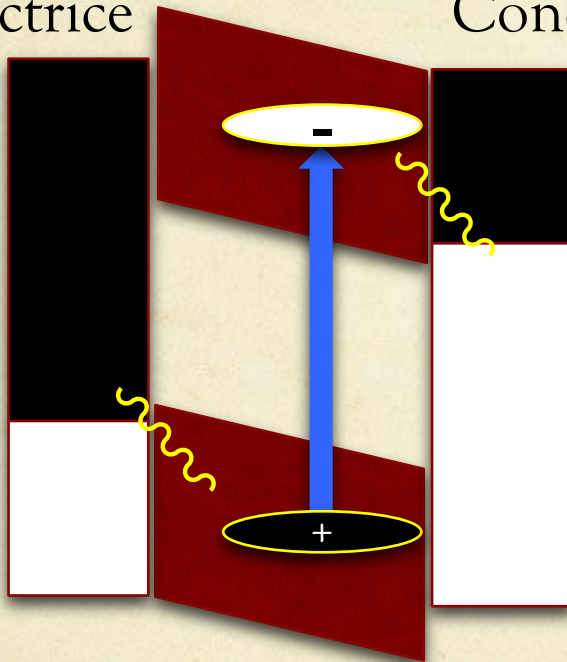


Photo-Voltaïque

Anode
Conductrice

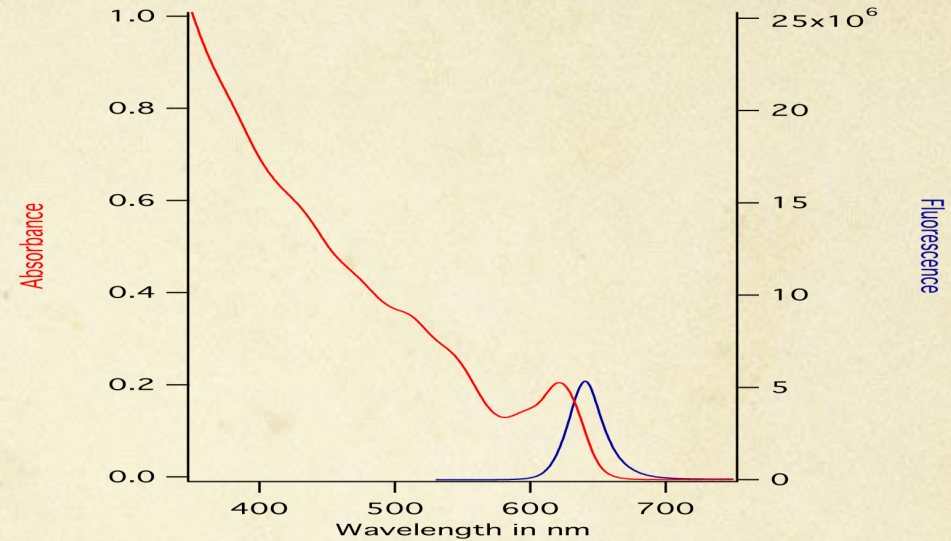
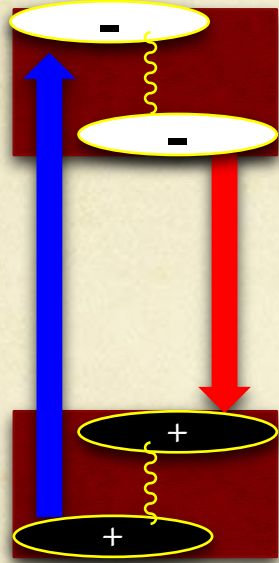
Cathode
Conductrice



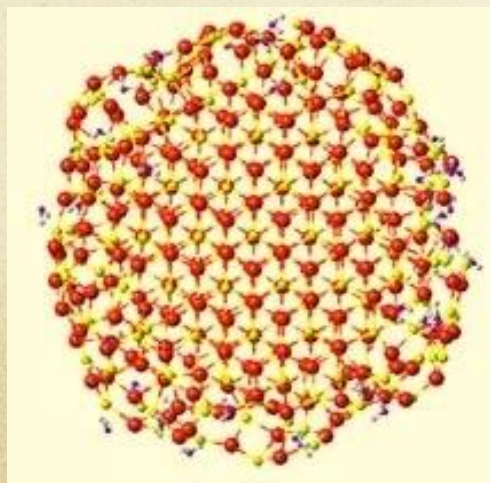
Semi-conducteurs

Bande de Conduction

Bande de Valence

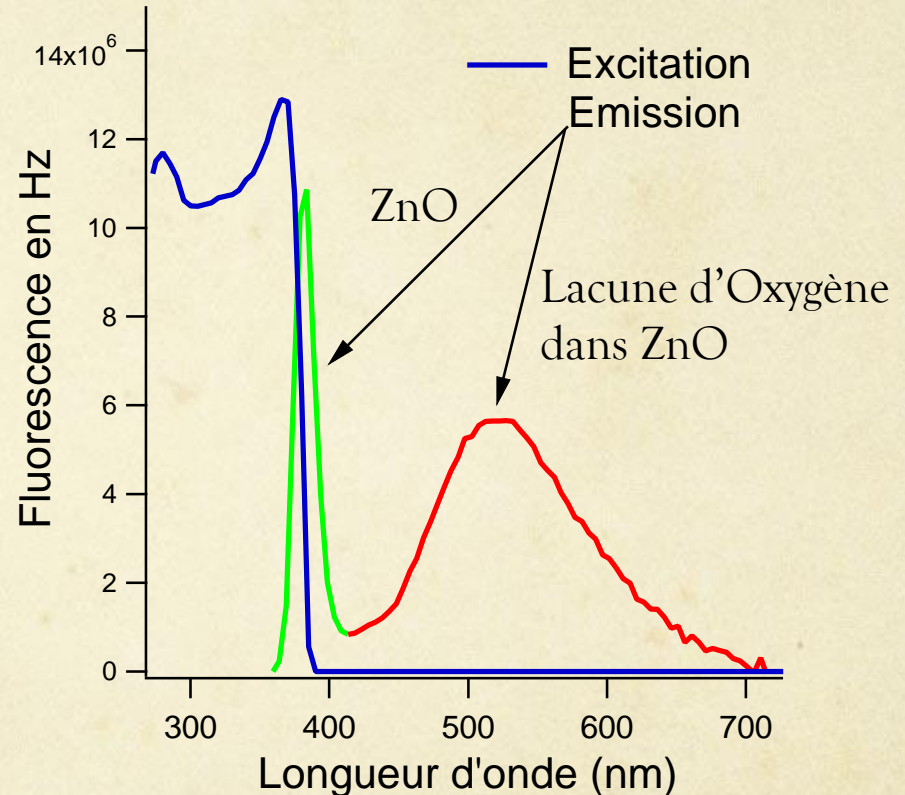
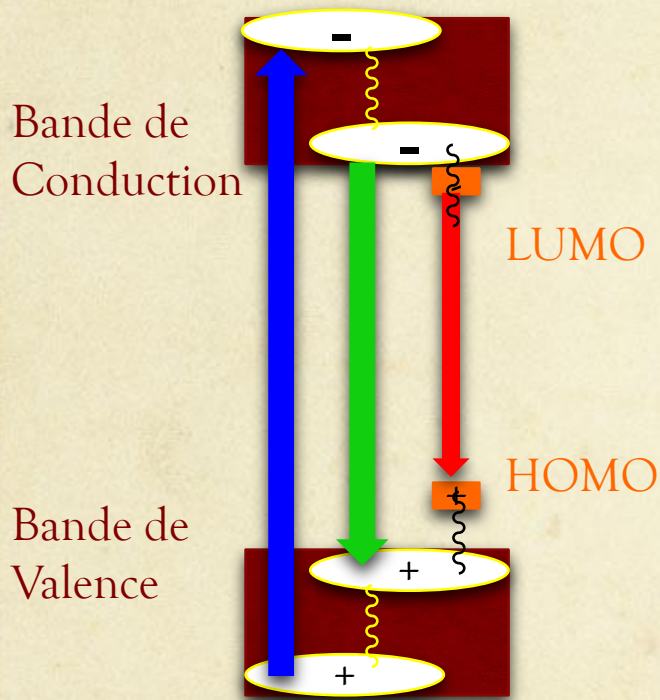


CdSe

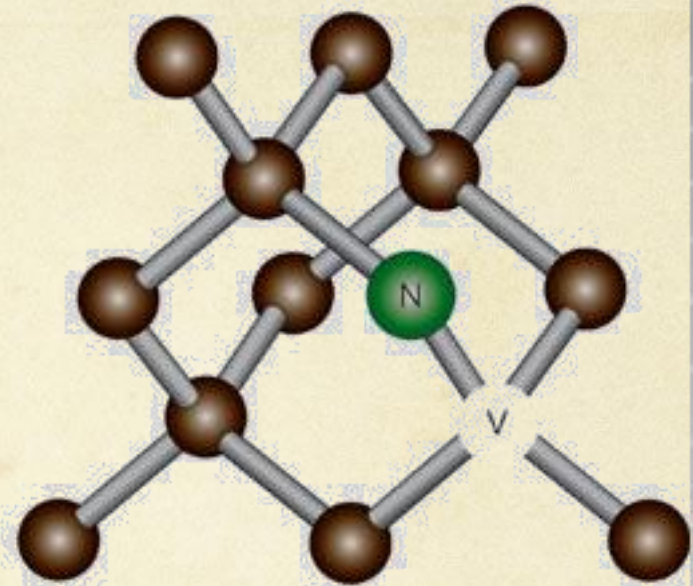
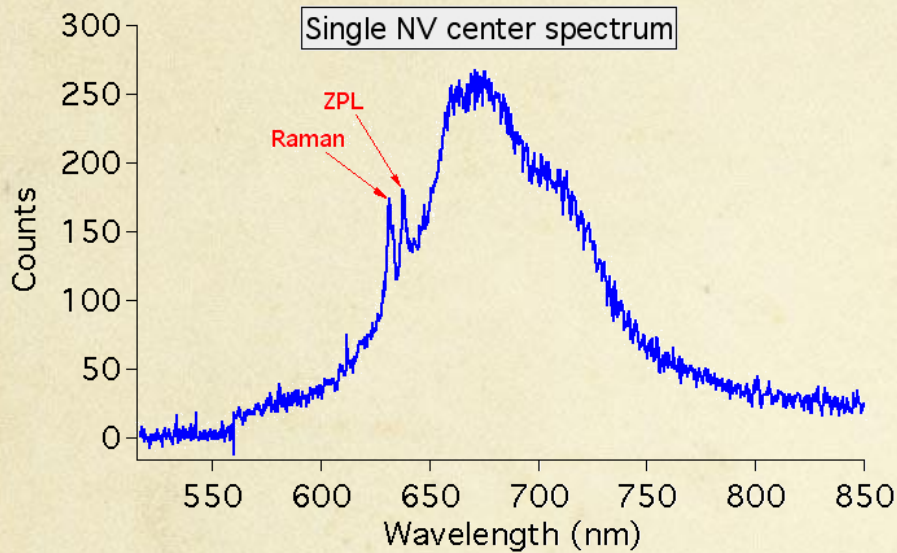


Size

Défauts dans les Semi-conducteurs



Impuretés dans le diamant

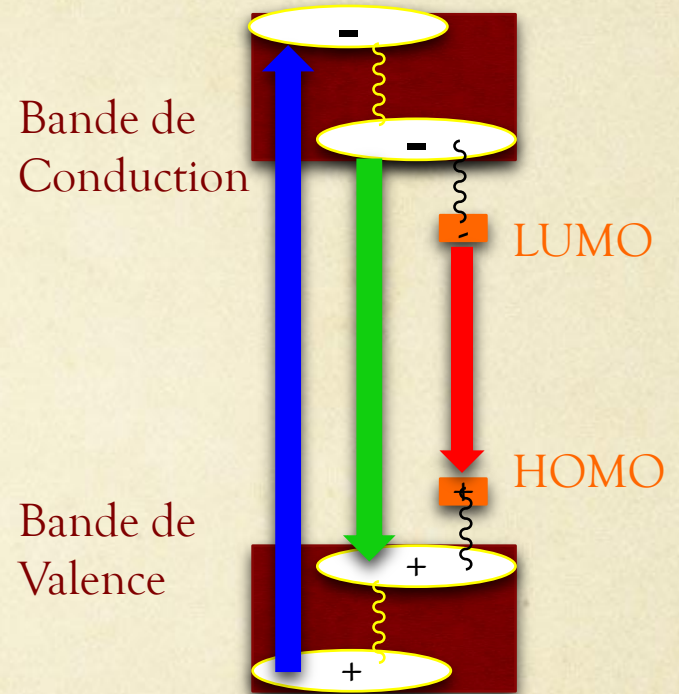
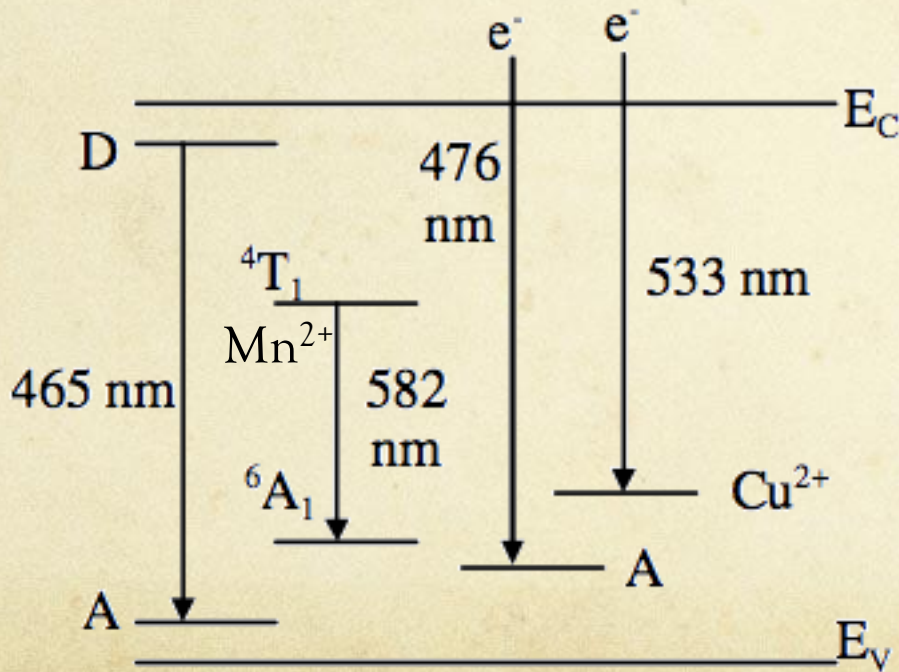


Spectroscopic and laser parameters of H3, NV and NE4 diamond

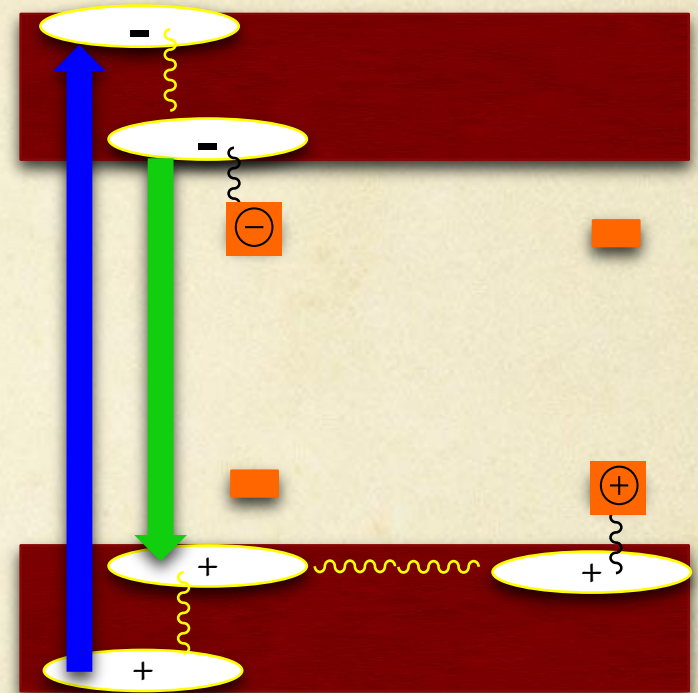
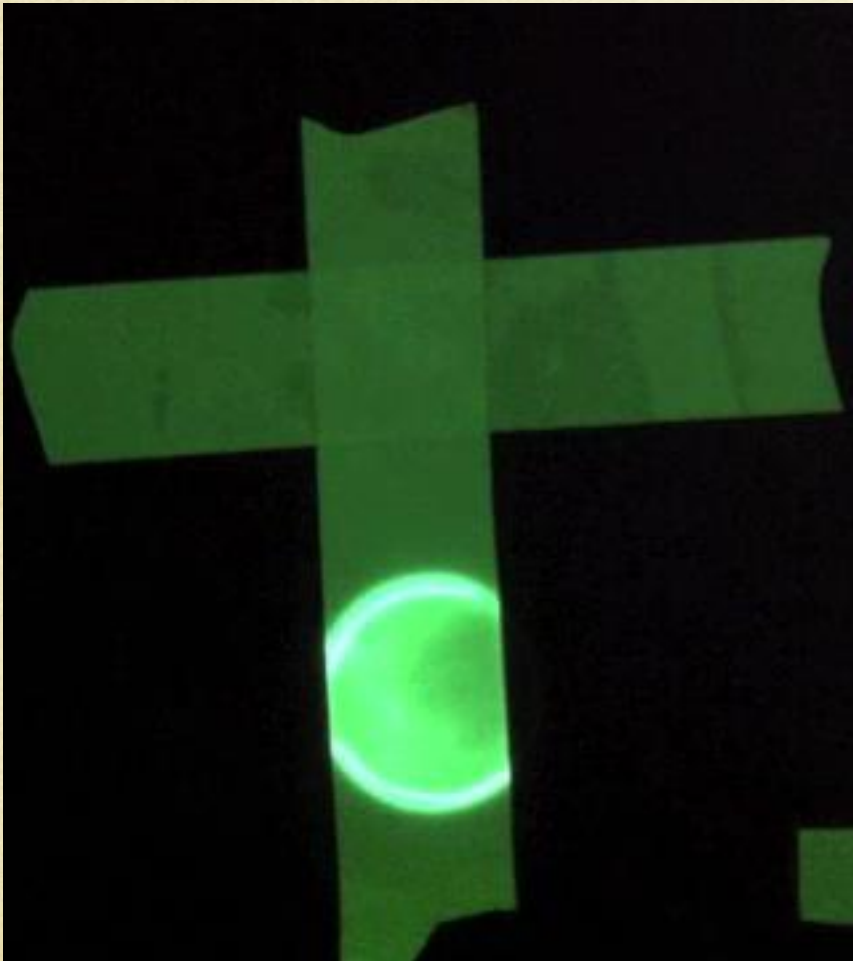
Color center	Absorption (nm)	Emission (nm)	Lifetime (ns)	Em. Width (nm)	Em. Cross- 1E-20 m ²
H3 V-N-V	490	525	16	60	1,6
NV N-V	565	690	13	120	3,2
NE4 V-Ni-V	670	950	27	100	6,2

Impureté du Semi Conducteur

$\text{Cu}^{2+}, \text{Mn}^{2+}, \text{Co}^{2+} / \text{ZnS}$



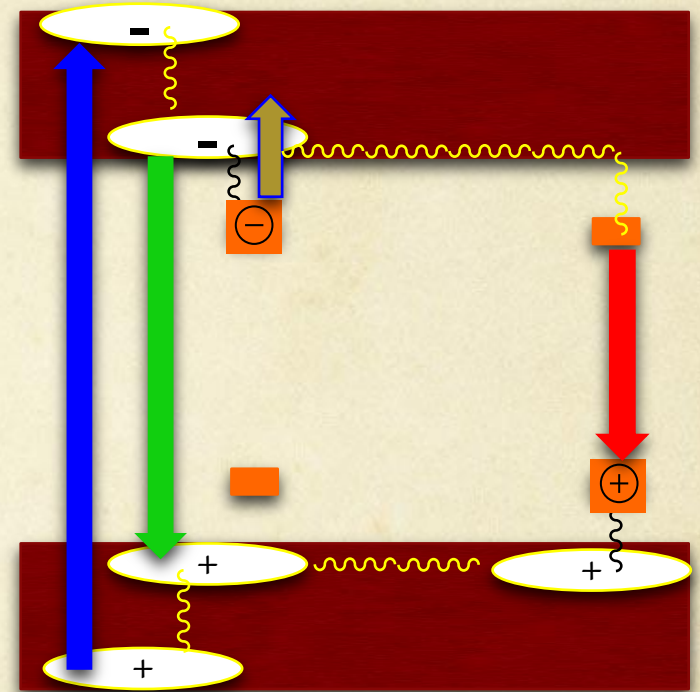
Thermoluminescence et Photo-dépiégeage



<http://www.culture.gouv.fr/culture/conservation/fr/methodes/thermolu.htm>

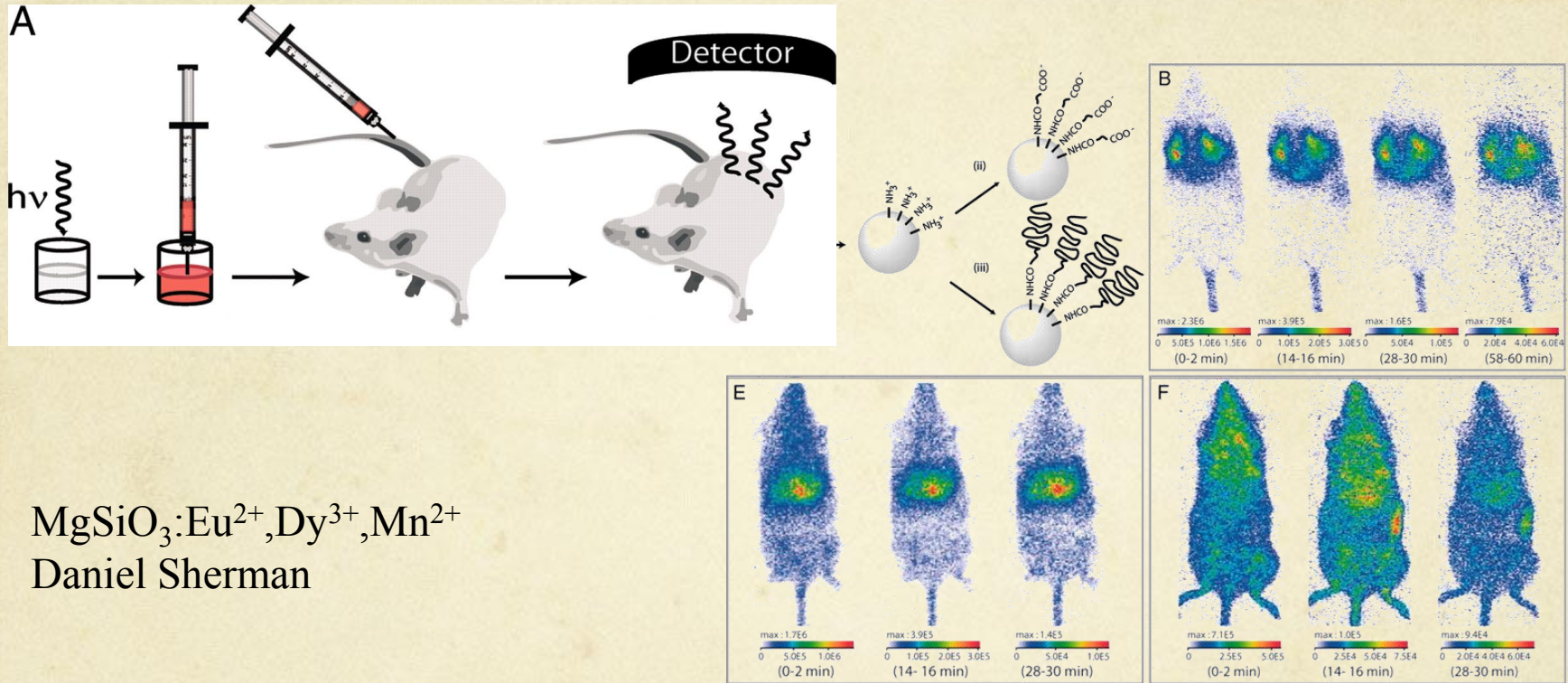
Schwankner, R. et al. J. Chemical Education 1981, 58, 806.

Thermoluminescence et Photo-dépiégeage



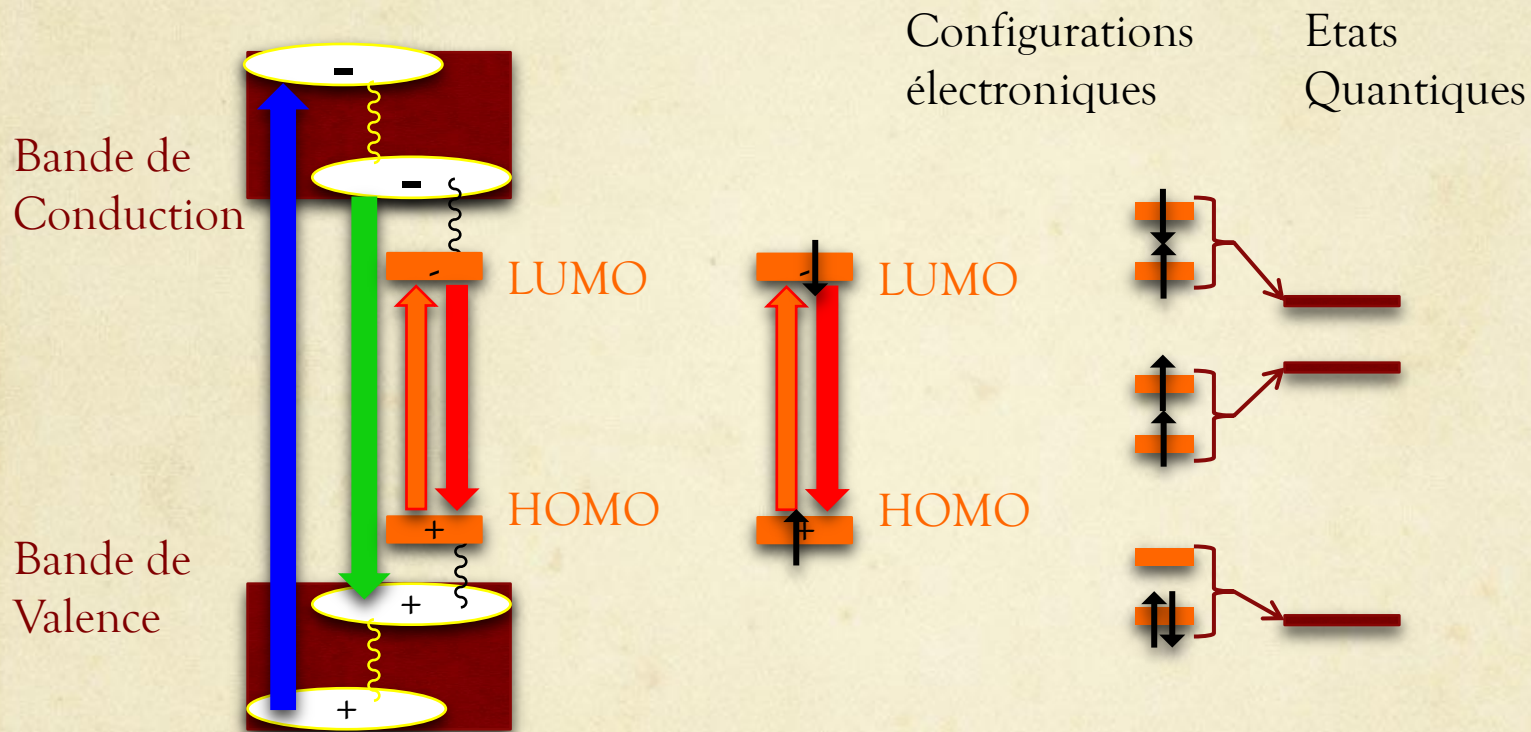
(1) Schwankner, R.; Eiswirth, M.; Venghaus, H. J. Chem. Ed. 1981, 58, 806.

Fonctionnalisation de surface et vectorisation

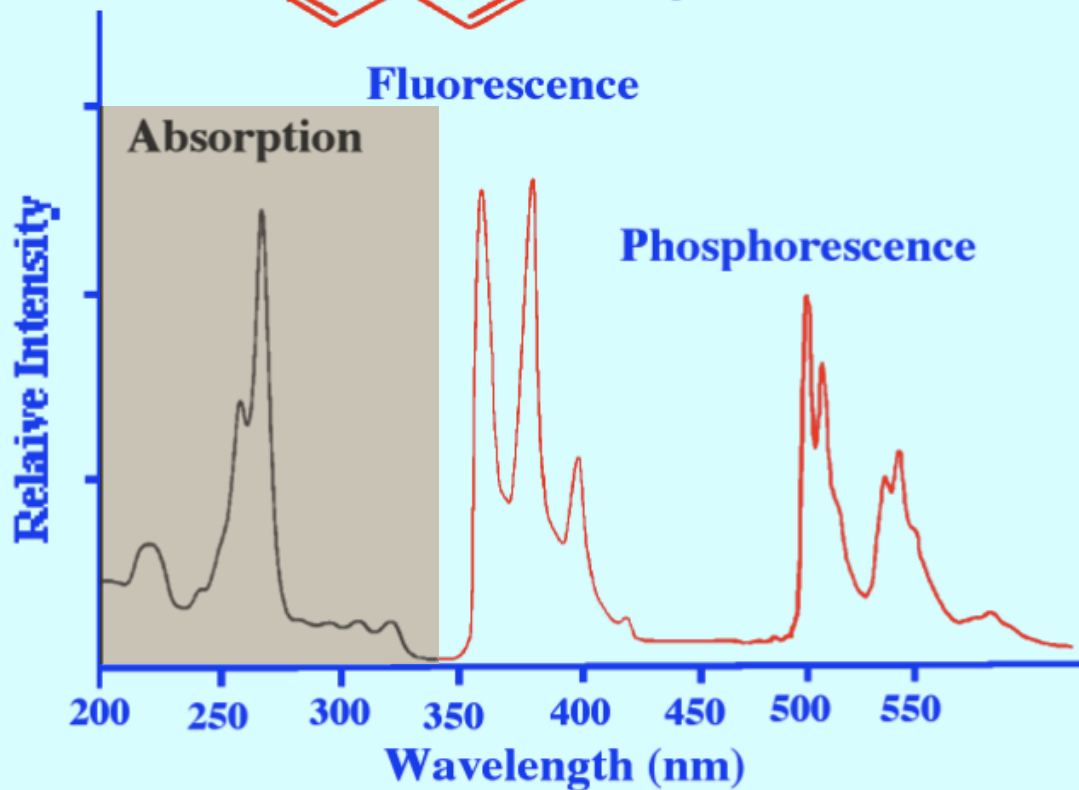
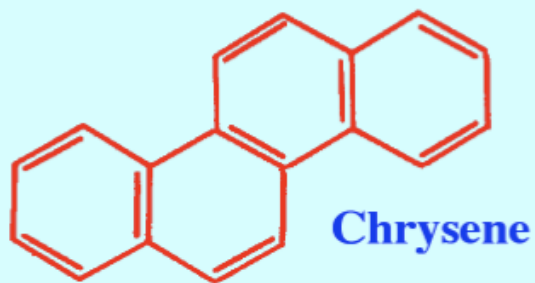


le Masne de Chermont Q et al. PNAS 2007;104:9266-9271

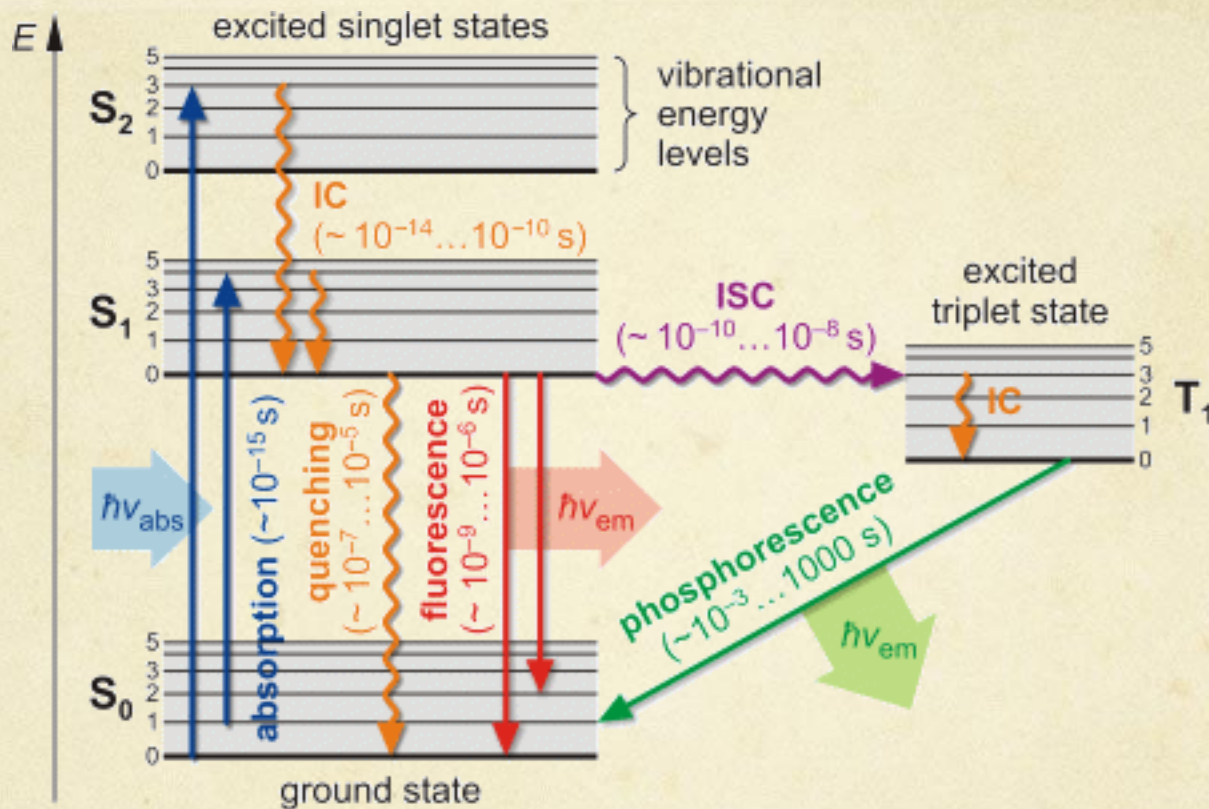
Fluorescence et Phosphorescence des molécules organiques



Fluorescence et Phosphorescence des molécules organiques

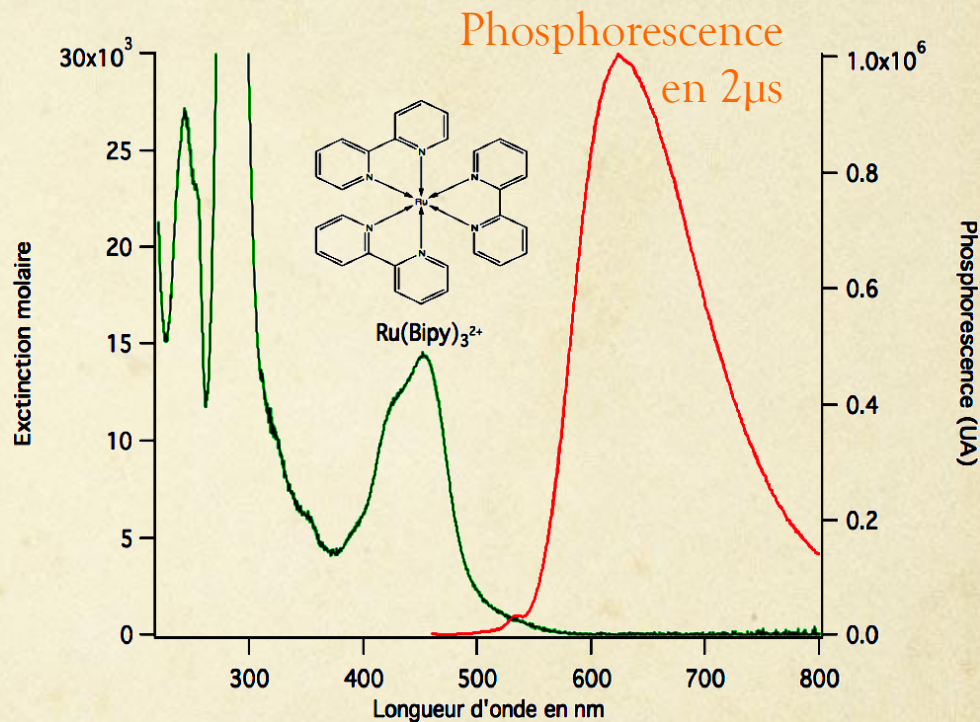
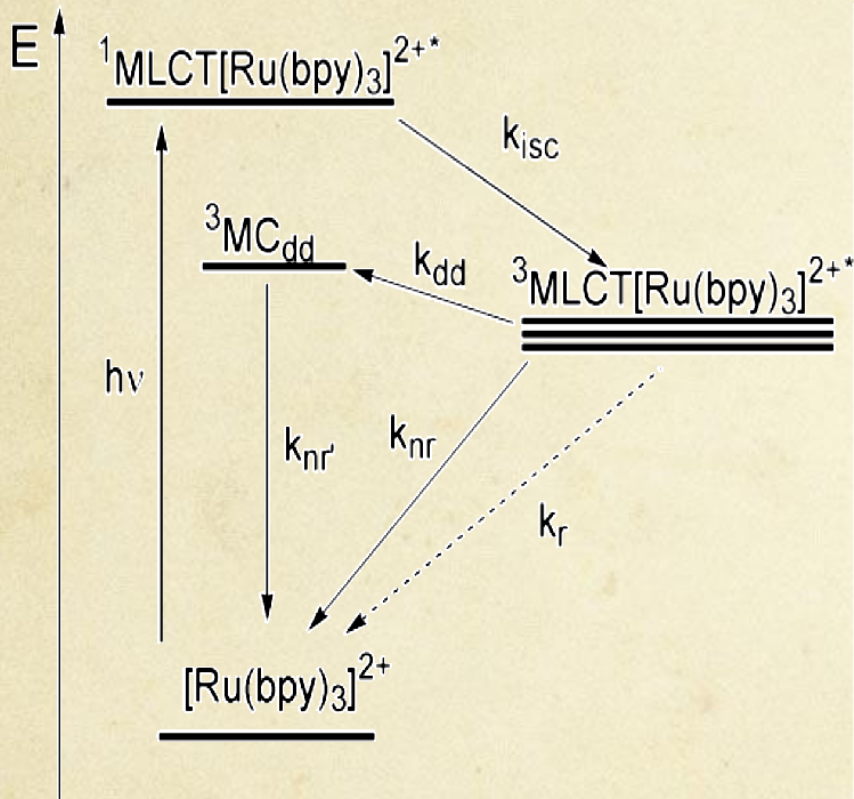


Fluorescence et Phosphorescence des molécules organiques



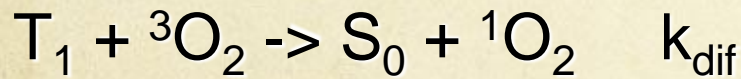
J.B. Birks. Photophysics of aromatic molecules. 1970
Drawing from diploma thesis of Steve Pawlizak, 2009.

Complexe de Ru⁺³ et Ir⁺³

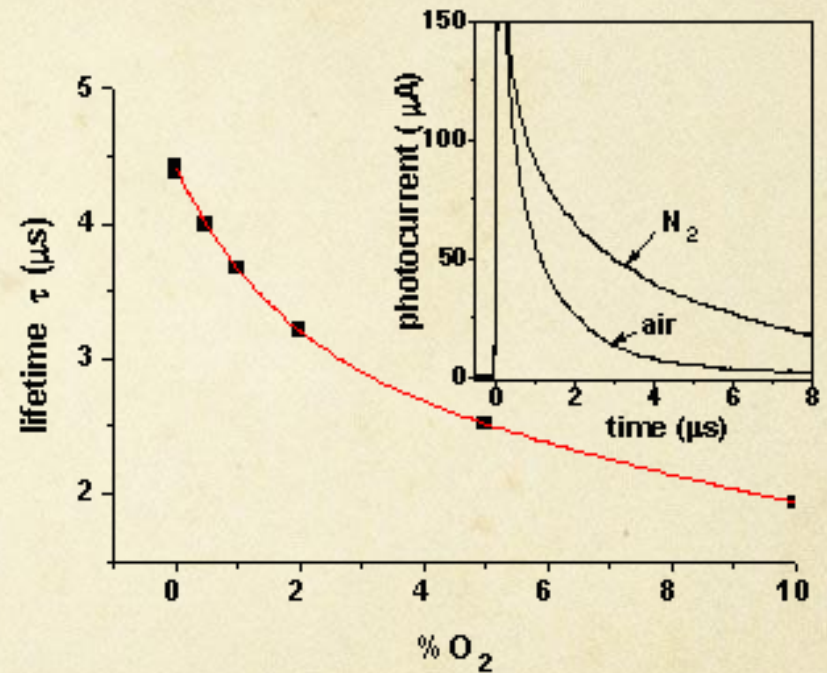


Maruszewski, K. et al. (1993) *JACS* 115 (18) 8345-50.

Détection O₂



$$\frac{d[T]}{dt} = - (k_{nr} + k_{phos})[T] - k_{dif} [O_2] [T]$$



Fluorescence, Phosphorescence et autres luminescences

la « Phosphorescence » de Cu,Mn,Co/ZnS

[http: IUPAC > Gold Book > alphabetical index > phosphorescence](http://IUPAC.org/GoldBook/alphabetic/index/phosphorescence) :

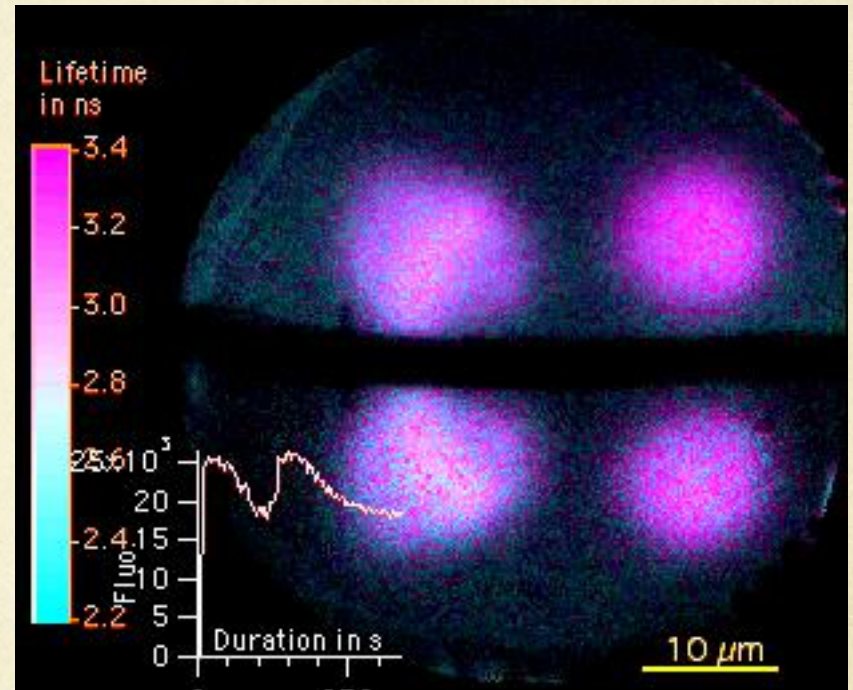
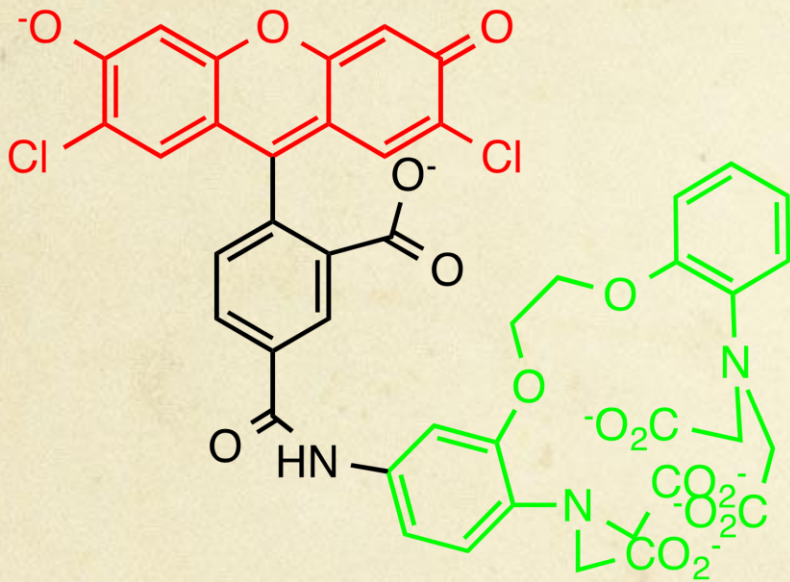
The term has been used to describe long-lived luminescence. The term designates luminescence involving change in spin multiplicity.

	Cu/ZnS	Mn/ZnS	Co/ZnS
Luminescence longue	☺	☺	☺
Non conservation du Spin	??	☺ ${}^4T_1 \rightarrow {}^6A_1$	☹ ${}^4T_1 \rightarrow {}^4A_2$
Thermo luminescence	☺	☺	
Photo dépiégeage	☺		
	??	Phosphore- scence Retardée	Fluorescence Retardée

Applications



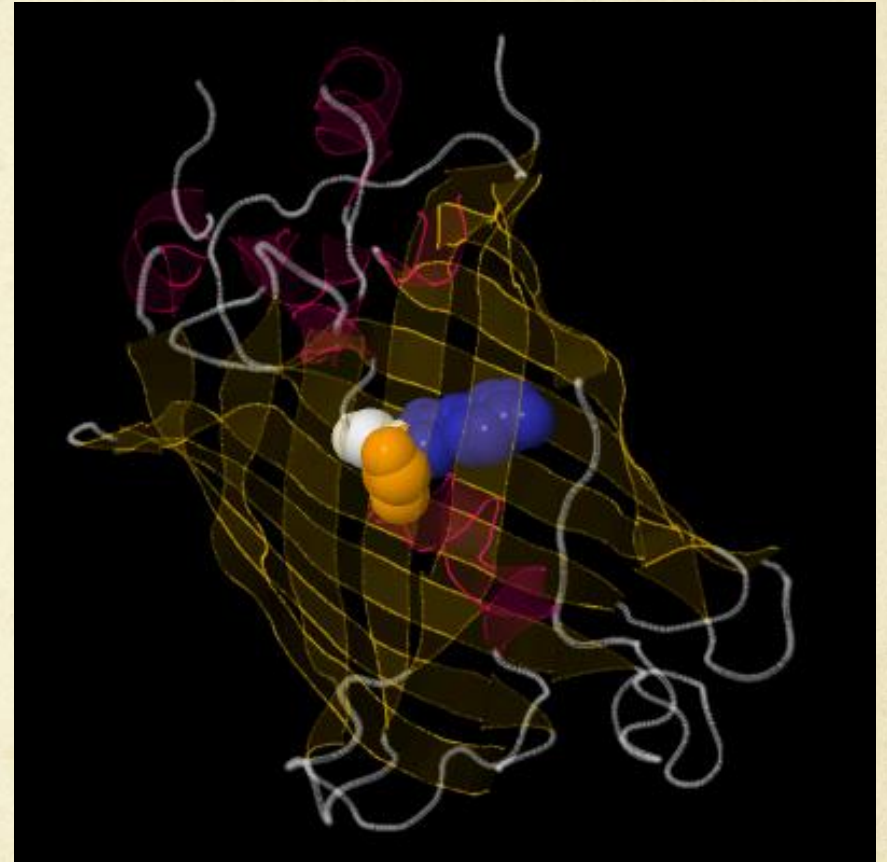
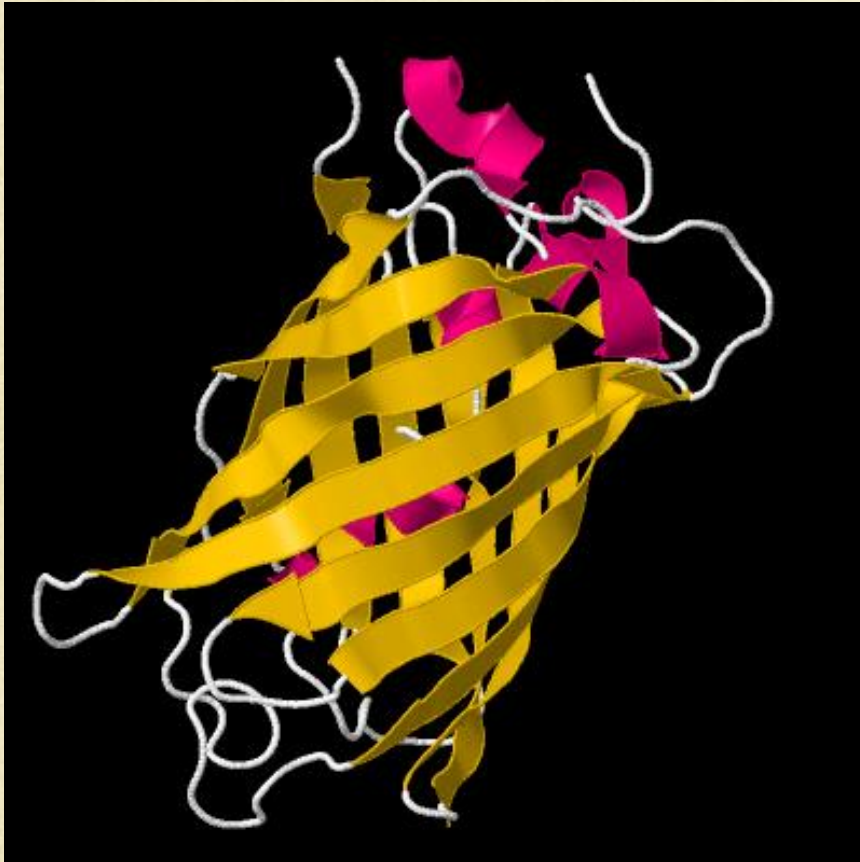
Indicateurs fluorescents



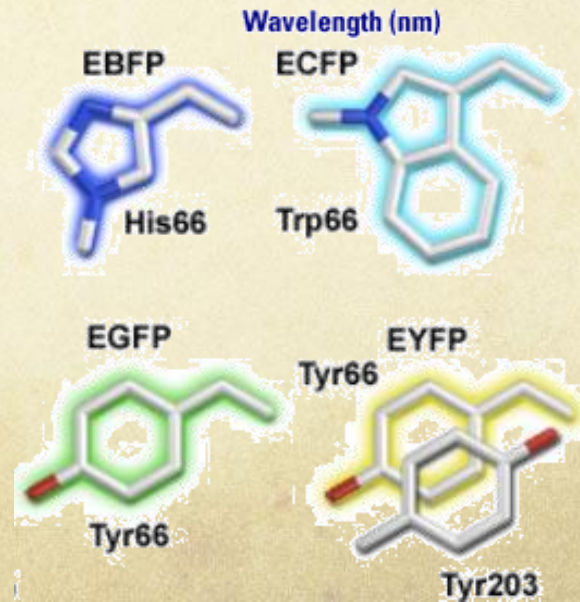
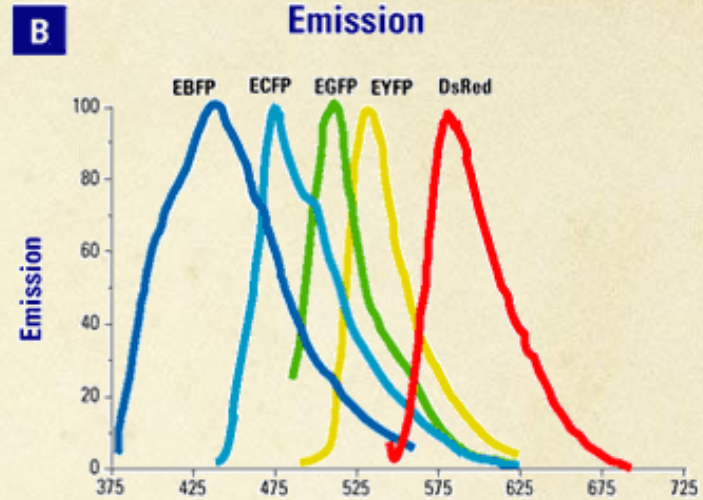
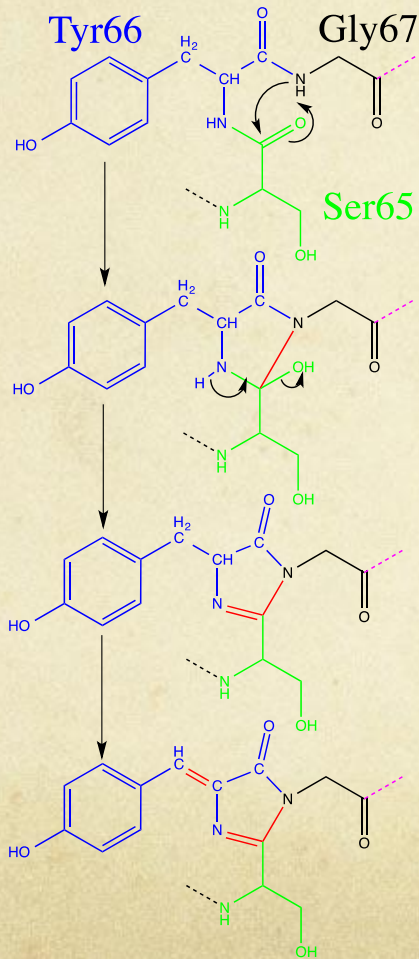
Tsien, R. Y. *Biochemistry* 1980, 19, 2396.

Valeur, B. <http://www.theses.fr/?q=%22Fluoroionophore%22>

Les protéines fluorescentes



Protéines fluorescentes



La super résolution

Photoactivated Localization Microscopy

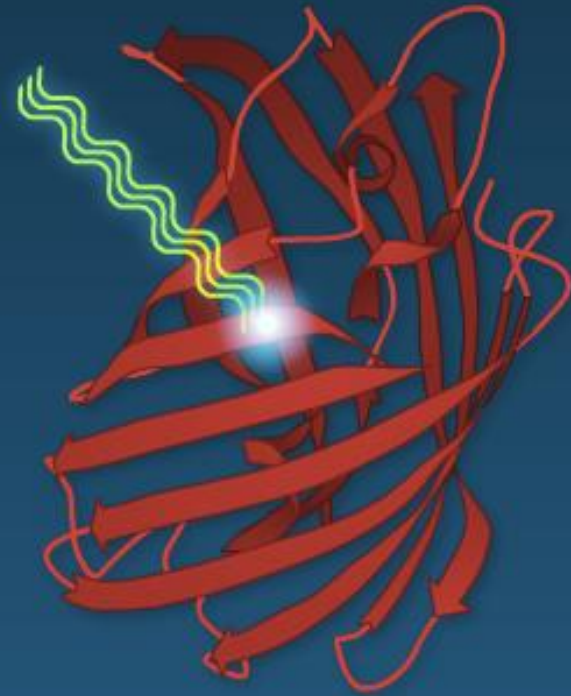
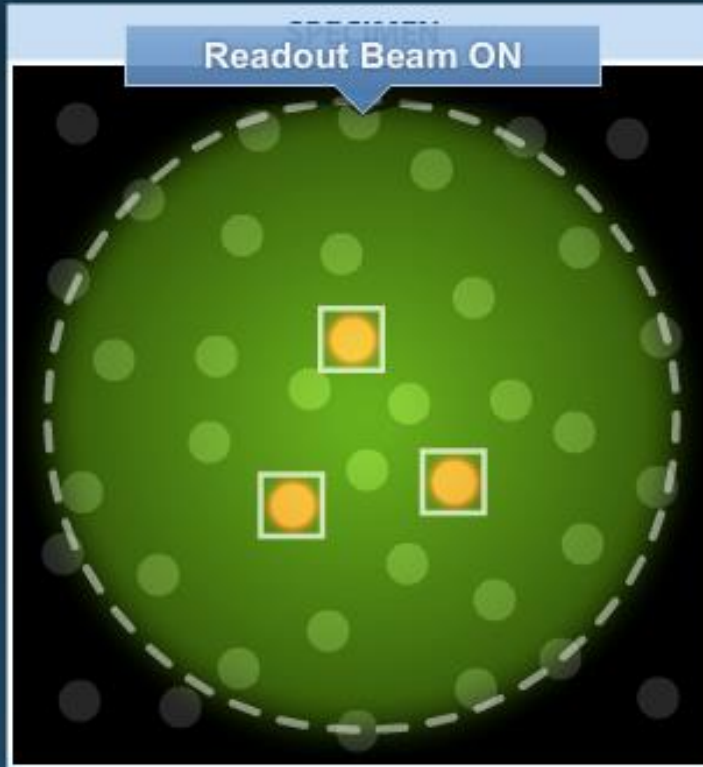


Image Activated Molecules

PALM Process



Choose a Specimen

Eos FP

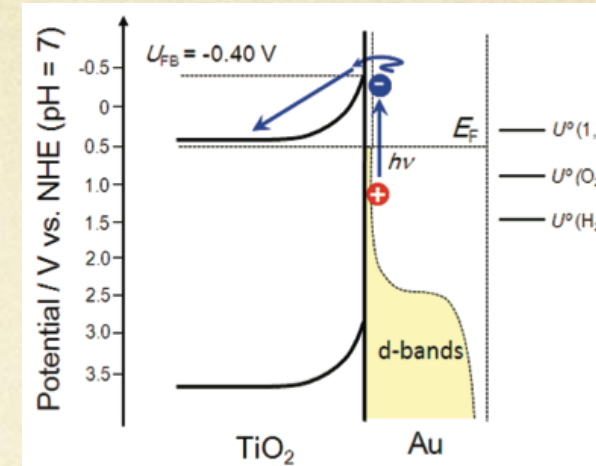
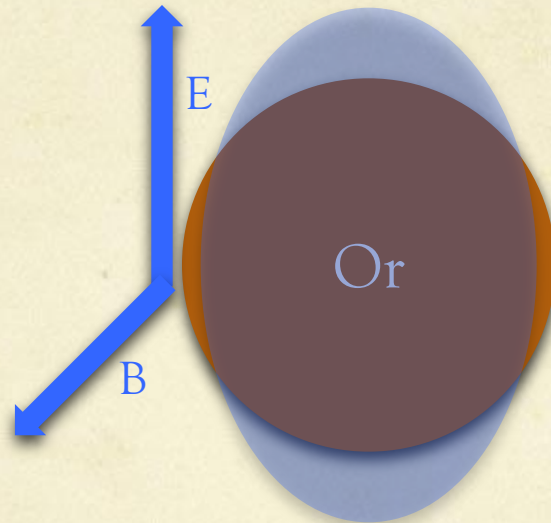
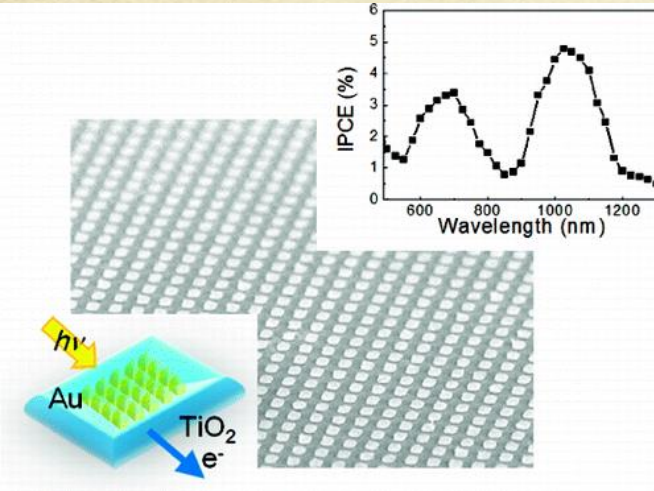


Conclusions

- Les diagrammes d'énergie permettent de décrire
 - Le déplacement de Stokes
 - Luminescence indépendante du mode d'excitation
 - Concept commun avec l'électronique
- De nombreuses applications
- Les prochains matériaux luminescents

Perspectives

○ Plasmon



Misawa, H. et al. J. Phys. Chem. Letters 2010, 1, 2031.

○ Thermo-Luminescence

Bibliographie

- Valeur, B. (2002) "Molecular fluorescence : principles and applications" Weinheim, Wiley-VCH 387 Pages.
- Werts, M. H. V. (2005) "Making sense of lanthanide luminescence" Science Progress 88 (Part 2) 101-31.
- Lakowicz, J. R. (2006) Principles of fluorescence spectroscopy Springer
- Cowan, R.D. (1981) Theory of Atomic Structure & Spectra, UCP, Berkley, p. 598-613 (Lanthanide/Actinide Configurations)
- Cox, P.A. (1987) The Electronic Structure and Chemistry of Solids, OUP, , p.137-145 (Lanthanides)
- West, A.R. (1984) [Solid State Chemistry](#), Wiley, London, Ch. 16 (Magnetism), Ch 17. (Optical Properties)
- Topics in Fluorescence Spectroscopy Lakowicz, Joseph R. (Eds.) chez Springer
- d

Distribution of populations

m: number of quenching sites, p: occupation probability, n: number of quenchers,

$$I_f(t) = I_f(0) \sum_{n=0}^m C_m^n p^n (1-p)^{m-n} \exp\left(-k_f t - n \int_0^t g(u) du\right)$$

$$I_f(t) = I_f(0) \exp(-k_f t) \left(p + (1-p) \exp\left(-\int_0^t g(u) du\right) \right)^m$$

$$\ln\left(\frac{I_f(t)}{I_f(0)}\right) \rightarrow -k_f t - m \int_0^t k_Q(u) du \quad \text{as } p \rightarrow 1$$

$$\ln\left(\frac{I_f(t)}{I_f(0)}\right) \rightarrow -k_f t + mp \left\{ \exp\left(-\int_0^t k_Q(u) du\right) - 1 \right\} \quad \text{as } p \rightarrow 0$$

33

Chimi- Bio- Luminescence

